Acta Crystallographica Section E Structure Reports Online ISSN 1600-5368 Editors: W. Clegg and D. G. Watson

μ -Chloro- μ -hydrido-bis[chloro(η^{6} -p-cymene)ruthenium(II)](Ru—Ru): a second polymorph

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metal-organic papers

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ISSN 1600-5368

Bruno Therrien,* Stéphane Frein and Georg Süss-Fink

Institut de Chimie, Université de Neuchâtel, Case postale 2, CH-2007 Neuchâtel, Switzerland

Correspondence e-mail: bruno.therrien@unine.ch

Key indicators

Single-crystal X-ray study T = 153 K Mean σ (C–C) = 0.004 Å R factor = 0.022 wR factor = 0.049 Data-to-parameter ratio = 18.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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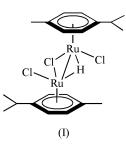
 μ -Chloro- μ -hydrido-bis[chloro(η^6 -p-cymene)ruthenium(II)](Ru—Ru): a second polymorph

The title dinuclear complex, $[Ru_2(C_{10}H_{14})Cl_4H]$, has twofold symmetry. The terminal Cl atom interacts with three C-H H atoms of symmetry-related neighbouring molecules. These interactions play a key role in the packing arrangement.

Received 4 October 2004 Accepted 13 October 2004 Online 22 October 2004

Comment

The title compound, (I), was first obtained by Bennett & Ennett (1984) from the reaction of $[(p-\text{Me-C}_6\text{H}_4-\text{Pr}^i)\text{RuCl}(\mu-\text{Cl})]_2$ with molecular hydrogen; the single-crystal X-ray structure analysis of (I), crystallized from a benzene solution, has been reported by Süss-Fink *et al.* (2000). In contrast with that crystal structure in the orthorhombic space group *Pbca*, we have analysed crystals of (I), which we had obtained from a dichloromethane solution. These crystals belong to the monoclinic space group *C*2/*c*, thus giving rise to a completely different packing arrangement from that in the crystals obtained from benzene. Although the molecular structure (bond lengths and angles) of (I) is almost identical to the earlier reported structure, the intermolecular interactions are quite different.



The dinuclear title complex, in which the bridging chloride and hydride ligands lie on a twofold axis, is shown in Fig. 1. The metal atom possesses one terminal and one bridging chloride; the bridging Ru–Cl distance is 2.4185 (14) Å, while the terminal Ru–Cl distance is 2.4021 (7) Å. The bridging Ru–Cl distance is shorter than that observed in other areneruthenium complexes possessing μ -Cl ligands (McCormick & Gleason, 1988; Gupta *et al.*, 1997; Therrien *et al.*, 1998; Bown & Bennett, 1999; Vieille-Petit *et al.*, 2002). The Ru–Ru distance of 2.9598 (7) Å is in accordance with a metal–metal single bond, and is comparable with the isoelectronic hydrido–trichloro complexes [(C₅Me₅)₂Rh₂Cl₂(μ -Cl)(μ -H)] (Churchill & Wei-Yang Ni, 1973) and [(C₅Me₅)₂Ir₂Cl₂(μ -Cl)(μ -H)] (Churchill & Julis, 1977).

Interestingly, the terminal Cl atom interacts with three C— H H atoms of two symmetry-related neighbouring molecules. These interactions are shown in Fig. 2 and summarized in Table 2. In the orthorhombic polymorph (Süss–Fink *et al.*, 2000), the terminal Cl atoms also interact with three C–H H atoms, but in that case with three symmetry-related neighbouring molecules. Furthermore, in the orthorhombic form, the dinuclear complex forms one-dimensional supramolecular chains through π -stacking interactions. Similar interactions are not observed in (I). Instead, the *p*-cymene moiety is surrounded by methyl groups. The closest C atoms are the isopropyl methyl C atoms, 3.77 and 4.31 Å from the benzene ring centroid of the *p*-cymene ligand.

Experimental

The compound was prepared as described by Bennett & Ennett (1984), and this monoclinic form was obtained by crystallization from dichloromethane solution.

 $D_x = 1.727 \text{ Mg m}^{-3}$

Cell parameters from 8000

 $0.36 \times 0.22 \times 0.14 \text{ mm}$

1749 reflections with $I > 2\sigma(I)$

Mo $K\alpha$ radiation

reflections

 $\mu = 1.72 \text{ mm}^{-1}$ T = 153 (2) K

Block, orange

 $R_{\rm int} = 0.040$

 $\theta_{\rm max} = 25.9^{\circ}$

 $h = -18 \rightarrow 18$

 $k = -21 \rightarrow 21$

 $l = -12 \rightarrow 12$

 $\theta = 2.0-25.9^{\circ}$

Crystal data

 $\begin{bmatrix} \text{Ru}_2(\text{C}_{10}\text{H}_{14})\text{Cl}_4\text{H} \end{bmatrix} \\ M_r = 577.92 \\ \text{Monoclinic, } C2/c \\ a = 14.782 \text{ (3) Å} \\ b = 17.431 \text{ (3) Å} \\ c = 10.425 \text{ (2) Å} \\ \beta = 124.16 \text{ (3)}^{\circ} \\ V = 2222.7 \text{ (11) Å}^3 \\ Z = 4 \\ \end{bmatrix}$

Data collection

Stoe IPDS diffractometer φ scans Absorption correction: multi-scan (Blessing, 1995) $T_{min} = 0.500, T_{max} = 0.786$ 8650 measured reflections 2163 independent reflections

Refinement

Refinement on F^2	H atoms treated by a mixture of
$R[F^2 > 2\sigma(F^2)] = 0.022$	independent and constrained
$wR(F^2) = 0.049$	refinement
S = 0.92	$w = 1/[\sigma^2 (F_o^2) + (0.0296P)^2]$
2163 reflections	where $P = (F_o^2 + 2F_c^2)/3$
119 parameters	$(\Delta/\sigma)_{\rm max} = 0.001$
	$\Delta \rho_{\rm max} = 0.44 \ {\rm e} \ {\rm \AA}^{-3}$
	$\Delta \rho_{\rm min} = -0.55 \ {\rm e} \ {\rm \AA}^{-3}$

Table 1

Selected geometric parameters (Å, °).

Cl1-Ru1	2.4185 (14)	Ru1-Ru1 ⁱ	2.9598 (7)
Cl2-Ru1	2.4021 (7)	Ru1-H1	1.711 (18)
Ru1 ⁱ -Cl2-Ru1	76.06 (3)	Cl2-Ru1-H1	82.1 (10)
Cl2-Ru1-Cl1	90.28 (2)	Cl1-Ru1-H1	87.80 (3)

Symmetry code: (i) $1 - x, y, \frac{3}{2} - z$.

The position of the hydride atom was derived from a Fourier difference map and refined, while the remaining H atoms were

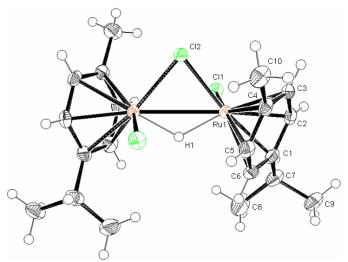


Figure 1

ORTEP-3 (Farrugia, 1997) drawing of (I), showing the atomic numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. Unlabelled atoms are related to labelled atoms by $(1 - x, y, \frac{3}{2} - z)$.

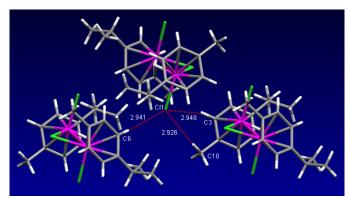


Figure 2

The intermolecular Cl···H interactions (dashed lines) observed in (I) (*MERCURY*; Bruno *et al.*, 2002).

Table 2

Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
C3-H3···Cl1 ⁱⁱ	0.93	2.95	3.758 (3)	146
$C10-H10B\cdots Cl1^{ii}$	0.96	2.93	3.770 (3)	147
$C6-H6\cdots Cl1^{iii}$	0.93	2.94	3.498 (3)	120

Symmetry codes: (ii) 1 - x, 2 - y, 2 - z; (iii) $x - \frac{1}{2}, \frac{3}{2} - y, z - \frac{1}{2}$.

included in calculated positions (C-H = 0.93–0.98 Å) and refined as riding, with $U_{iso}(H) = 1.2$ or 1.5 times U_{eq} (parent atom).

Data collection: *EXPOSE* in *IPDS Software* (Stoe & Cie, 2000); cell refinement: *CELL* in *IPDS Software*; data reduction: *INTE-GRATE* in *IPDS Software*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

This work was supported by the Swiss National Science Foundation (grant No. 20-61227-00). We thank Professor H.

Stoeckli-Evans for helpful discussions and access to X-ray facilities.

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